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By William B. Motherwell

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Radical (chemistry) - Wikipedia, the free -

Chain reactions involving free radicals can usually be divided into three distinct processes. These are initiation, propagation, and termination.

http://en.wikipedia.org/wiki/Radical_reaction

Section 17.2: Radical chain reactions - Chemwiki -

17.2A: The three phases of radical chain reactions. Because of their high reactivity, free radicals have the potential to be both extremely powerful chemical tools

http://chemwiki.ucdavis.edu/Organic_Chemistry/Organic_Chemistry_With_a_Biological_Emphasis/Chapter_17:_Radical_reactions/Section_17.2:_Radical_chain_reactions

www.lib.uoi.gr -

Practical Organic Synthesis: Name Reactions A Collection of Detailed Reaction Mechanisms. Modern Oxidation Methods. J.-E. Backvall, Wiley-VCH.

http://www.lib.uoi.gr/services/book_delivery_2006.php?page=900&journal=

Do alpha-acyloxy and alpha-alkoxycarbonyloxy -

Most organic radical reactions occur through a it is so slow as to have limited synthetic value in radical chain Radical Reactions in Organic Synthesis.

<http://www.ncbi.nlm.nih.gov/pmc/articles/PMC1524793/>

David Crich (Foreword of Modern Synthetic Methods -

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https://www.organicdivision.org/nos_history

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Substitution Reactions - Free Radical Chain -

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<http://www.sciencedirect.com/science/article/pii/B9780080924953500106>

Titanocene(III) chloride mediated radical-induced -

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http://www.academia.edu/2390928/Titanocene_III_chloride_mediated_radical-induced_addition_to_Baylis-Hillman_adducts_Synthesis_of_E_-and_Z_-trisubstituted_alkenes_and_%CE%B1-Methylene_Arylidene_%CE%B4-Lactones

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this article focuses on recent advances in the field of organic synthesis with of new synthetic reactions, methods radical reactions and total synthesis.

<http://rspa.royalsocietypublishing.org/content/470/2163/20130690>

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http://www.ucl.ac.uk/chemistry/staff/academic_pages/william_motherwell/publications

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